WARNING

You must return this paper with your answerbook, otherwise marks will be lost.

EXAMINATION NUMBER

AN ROINN OIDEACHAIS **JUNIOR CERTIFICATE EXAMINATION, 1993**

SCIENCE - ORDINARY LEVEL

TUESDAY, JUNE 15 - Afternoon 2.00 - 4.30 p.m.

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SECTION A TO BE ANSWERED BY ALL CANDIDATES. Sections B, C, D, E are on separate sheets. Answer the questions in the spaces provided.

SECTION A - CORE (144 marks)

Answer any 12 parts (a), (b) (c) etc. from this section.

- Return this Section of the examination paper in your answer book. Name the pieces of apparatus shown in the drawings: (a) (b) The drawing shows a ray of white light passing through a piece of glass A. What is A called? What effect does A have on the beam of light? Name the primary light colours Heat is transferred by: (c) CONDUCTION CONVECTION or RADIATION In which way is heat transferred in a liquid?
 - In which way is heat carried from the Sun to the Earth? Name one method used to prevent heat loss in our homes.

OVER->

(d)	The WATT is a unit used in electricity. What does the watt measure?	WIRE [
	The drawing shows a 5 A fuse. What happens if the current going through the fuse is more than 5 amperes?	
	An electric fire uses two units of electricity every hour. Electricity costs 8 p per unit. How much does it cost to use the fire for 5 hours?	
(e)	The universe is made of many galaxies. What is the name of the galaxy to which the Earth belongs?	
	What is the solar system?	
	How long does it take the Earth to orbit the Sun ?	
(f)	Bottles in the laboratory may carry warning (hazard) symbols. What does the symbol shown tell you ?	
	Choose words from the following MELTING FREEZING EVAPORATION CONDENSA to describe the changes in (i) and (ii) below: (i) Liquid → Solid	TION
	(ii) Gas → Liquid	
(g)	What is a compound?	
	The drawing shows an insect (pond skater) which can walk on water. What property of water allows an insect to walk on water?	

OVER->

)	Name the gas which makes up about 20% of the air.	<u>-</u>
	Air was sucked through the flask as shown. The lime water turned milky. What does this tell you about air ?	AIR →
	What gas in the air is needed for burning ?	-
	A piece of pink cobalt chloride paper was left in air. It turned blue. What does this tell you about air ?	
	Sand is INSOLUBLE in water. What does the word INSOLUBLE mean ?	_
	Sugar dissolves in the water to form a sugar solution. The water is called a SOLVENT. What does the word SOLVENT mean ?	-
	Atoms are made up of small particles. Name the <i>three</i> types of particles.	
	Which one of these particles is involved in chemical reactions?	-
	Give two CHARACTERISTICS of living things. (I)	_
	(ii)	_
	Give TWO differences between animals and plants.	
	(a)(b)	











POTATO

CHEESE

APPLE

MILK

BREAD

Cho	ose one food from the drawings above which has a large amount of:		
(a)	Carbohydrate		
(b)	Calcium		
Wha	it is the function of fat in the human body ?		
Wha	it is the function of protein in the human body ?		
	habitat there are plants and animals. se ONE way in which the animals compete with each other.		
Give	e an example of how animals and plants help each other (co-operate).		
— Nan	ne the organ or part of the human body which does each of the following:		
	nps blood around the body.		
•	ses oxygen into the body.	-	
give	s the body its shape and ps it upright.		
mai	ces egg cells in the female.	=	
			BACTE
Bac	teria are examples of micro-organisms.	29	
Wha	diagram shows bacteria growing on a dish. at substance needs to be added to the dish ore bacteria will grow ?		
Nar	ne a disease caused by a virus.		
	ne a microorganism used in the		

11

JUNIOR CERTIFICATE EXAMINATION, 1993

SCIENCE - ORDINARY LEVEL (N.B. Not for Science - Local Studies candidates)

TUESDAY, JUNE 15

Section A is on a separate sheet which provides spaces for your answers.

The completed sheet should be enclosed in your answer-book.

SECTIONS B, C, D, E.

The questions from these sections should be answered in your answer-book.

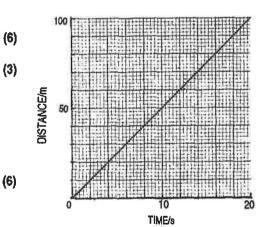
Choose any three sections from B, C, D, E.

Answer two questions from each chosen section. All questions carry equal marks.

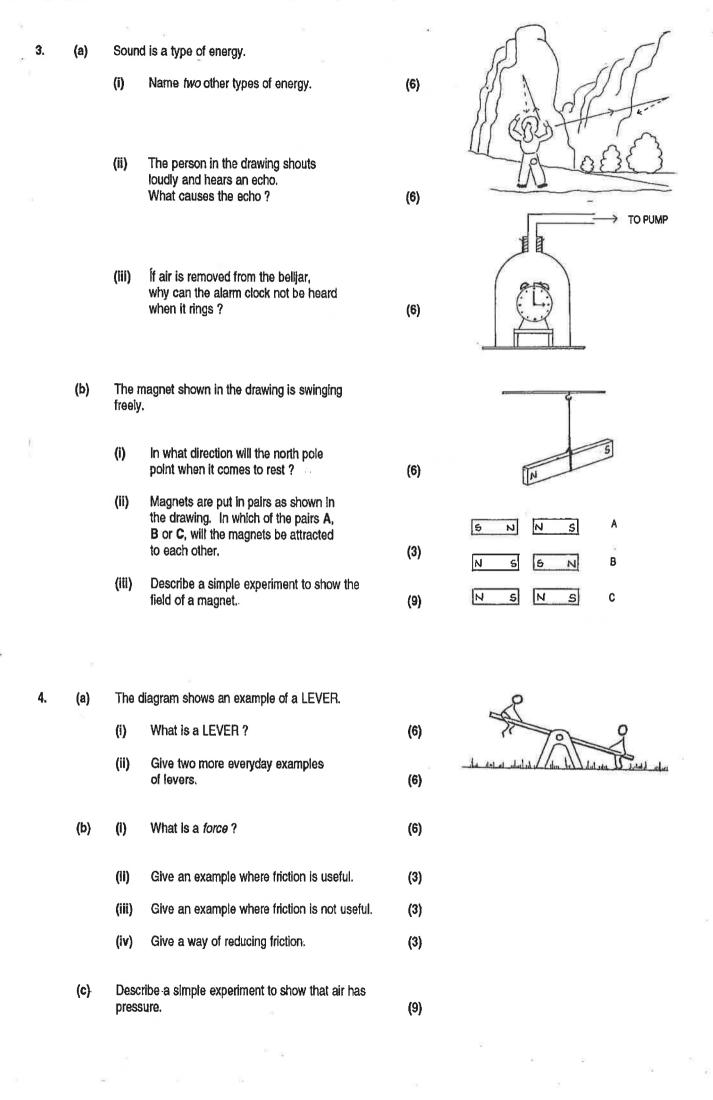
SECTION B PHYSICS (72 marks)

Answer any two questions

- 2. (a) (i) What piece of apparatus would you use to measure the mass of a stone? (3)
 - (ii) Describe an experiment you could do to measure the volume of the stone. (9)
 - (iii) If the mass of stone is 75 g and its volume is 25 cm³, what is the density of the stone? (9
 - (b) (l) What is speed?
 - (ii) What are the units of speed?
 - (iii) A cyclist travels 100 metres in 20 seconds as shown on the graph. What is the average speed of the cyclist?



OVER->



SECTION C CHEMISTRY (72 marks)

Answer any two questions.

- (a) (i) What is an element? (6)(ii) Name the elements which have the symbols:
 - Na Fe (6)
 - (III) The periodic table arranges the elements in groups. Group 7 is shown in the drawing.
 Why are these elements put in the same group? (6)

9
19.00
17 Cl 35.457
35 Br 79.916
53 I 126.92

- (b) (i) Write down a word equation for the reaction between hydrochloric acid and calclum carbonate. (6)
 - (ii) Write a word equation for the reaction of magnesium and oxygen. (6)
 - (III) During a chemical reaction an ATOM can be changed into an ION?
 What is an ION? (6)

- 6. (a) (i) Name two everyday metals. (6)
 - (ii) Give one use for each of the metals you have named. (6)
 - (iii) Metals are MALLEABLE and DUCTILE Explain what ONE of these words mean. (6)
 - (b) Metals are conductors of heat.

 Describe a simple experiment to show that a metal is a good conductor of heat. (9)
 - (c) Sometimes we use metals in the form of an ALLOY.
 What is an alloy? Name an alloy. (9)

- 7. (a) An INDICATOR is a chemical which changes colour when it is put into an acid or base.
 - (i) Name an indicator you have used in the laboratory.

(3)

(ii) Describe a simple experiment to obtain an indicator from red cabbage.

(9)

(b) The pH scale is used to compare the strength of acids and bases. Here is a list of liquids with their pH values:

LIQUID	A	В	c	D	E
pH	1	5	7	9	12

(i) Which liquid is the strongest acid?

(3)

(ii) Which liquid is the strongest base?

(3)

(iii) Which liquid is neutral?

(3)

(iv) What would you use to measure the pH of a liquid?

(3)

(c) (i) The drawing shows dilute hydrochloric acid reacting with the metal magnesium.

The gas given off burns with a pop.

What is the name of the gas?

(6)

(ii) Acids can dissolve in rain to form acid rain.
Give TWO harmful effects of acid rain.

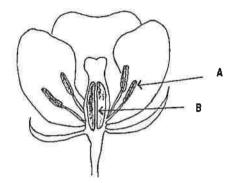
(6)



SECTION D BIOLOGY (72 marks)

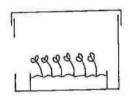
Answer any two questions

- 8. (a) (I) Name parts A and B of the flower shown in the diagram.
 - (6)
 - (ii) What is the function of part A?
- (6)
- (iii) What is the function of part B?
- (6)



Seeds were put on damp cotton wool in a box which had an opening on one side.
 The seedlings grew towards the light as shown in diagram.
 What is the name given to this response?





(c) Describe a simple experiment to show the movement of water in a plant.

(12)

- 9. (a) The drawing shows a model of the breathing system in humans.
 - (i) What do the balloons represent (stand for)?
- (3)
- (ii) What does the rubber represent (stand for)?
- (3)
- (iii) What happens when the rubber sheet is pulled down?

(6)

- (b) (i) Explain what the word EXCRETION means. (6)
 - (6)
 - (ii) The drawing shows one of the excretory systems in humans.Name the parts marked A and B.



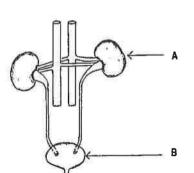
(c) (i) Name one mineral which makes bones hard.

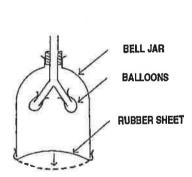


(9)

(6)

(ii) Describe a simple experiment to show what happens to bones when the mineral you have named in (i) is taken out.

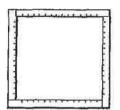




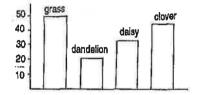
- 10. (a) (i) The drawing shows a piece of apparatus which is used to investigate the number of different plants in a habitat.

 Name the apparatus.
- (6)

(3)



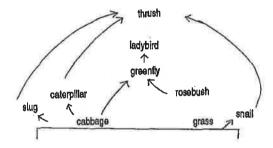
- (ii) The drawing shows the result of a survey of plants growing in a habitat.
 - Which type of plant occurred most often ?
 - Which type of plant occurred least often? (3)



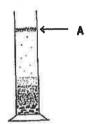
- (b) Some living things are PRODUCERS and some are CONSUMERS.
 - (i) Name TWO consumers shown in the diagram.
 - (ii) What is a PRODUCER?

(6)

(6)



- (c) (i) The drawing shows some soil which has been shaken up with water and allowed to stand.
 - Name the part which is found at A.
- (3)
- (ii) Describe a simple experiment to find the mass of water in a sample of soil. (9)



SECTION E APPLIED SCIENCE

Answer any two questions

11. EARTH SCIENCE

- (a) (i) Name ONE planet (other than Earth) which orbits the Sun. (3)

 (ii) Distances to stars are often measured in light years. What is a light year? (6)
 - (iii) Explain how an eclipse of the sun occurs. (9)
- (b) In the water cycle, water evaporates all the time from the surface of Earth. Describe a simple experiment to show the effect of temperature on the rate of evaporation. (12)
- (c) Name two measurements you would take if you were weather recording. (6)

12. HORTICULTURE

- (a) (i) Name two living organisms you would find in soil. (6)
 - (ii) Explain how ONE of the living organisms you have named in (i) is useful in the soil. (6)
- (b) Describe how you could take softwood or hardwood cuttings from a plant and make them grow. (12)
- (c) (i) Describe how you would measure the germination rate of seeds. (9)
 - (ii) What is DORMANCY? (3)

MATERIAL SCIENCE 13. Use words from the list below to answer this question. (a) (Choose FOUR words only) POLYTHENE COAL COPPER OAK GLASS WOOL Name a metal. (i) (11) Name a textile. (iii) Name a plastic. Name a timber. (4×3) (iv) (b) (i) Give one use of a plastic in your home. (3) (ii) Give one use of wood in your home. (3) (c) Answer one of the following questions: A, B, C or D. A: PLASTICS Give one advantage and one disadvantage of (i) using plastics. (6) (ii) Describe a simple experiment to compare the hardness of two different plastics. (12)OR **B: TEXTILES** Name two man-made (synthetic) fibres. (i) (6) (ii) Describe a simple experiment to compare the value of two textiles as insulators. (12)OR C: METALS (l) Name a metal which is found in a pure state in the ground. Name a metal which is found as an ore. (6)(II) Describe a simple experiment to compare the reactivity of two metals. (12)

OR

Name two types of man-made or manufactured

Describe a simple experiment to compare the

bending strength of two pleces of wood.

(6)

(12)

D: TIMBER

timbers.

(i)

(ii)

14. FOOD

- (a) (i) What chemical would you use to test a piece of bread for starch?
- (3)
- (ii) What effect does this test have on the colour of the bread?
- (3)

(6)

(12)

(6)

(6)

(9)

(iii) The information shown in the chart was printed on a packet of breakfast cereal.

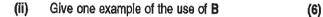
How much energy is there in 200 g of the cereal.

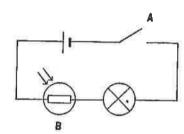
100 grams provides: Protein 6 g Vitamin A 1.8 mg Sodium 1.0 g Iron 6.7 mg Energy 1650 kJ

- (b) Describe a simple experiment to make silage.
- (c) (i) Name two methods of food preservation for human use.
 - (ii) Explain why one of the methods you have named preserves food.

15. ELECTRONICS

- (a) (i) Name the parts marked A and B in the circuit.





- (b) Study the circuit diagram shown.
 - (I) Name the device labelled A.

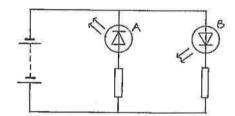
(3)

(ii) Through which of the devices (A or B) will a current flow?

(3)

(III) Explain why the current flows as you have stated.

(3)



(c) (i) Draw a simple circuit diagram containing a two-way switch and a device to show when a current is flowing.

Label your diagram.

(9)

(Ii) Draw the symbol for a variable resistor.

(3)

16. ENERGY CONVERSIONS

- (a) Give an example of each of the following energy conversions:
 - (i) Potential energy to kinetic energy.
- (3)
- (ii) Light energy to electrical energy.
- (3)
- (iii) Describe a simple experiment to show that chemical energy can be converted to heat energy.
- (6)
- (b) The drawing shows a simple electromagnet.
 - (i) What is the iron inside the coil called?
- (3)
- (ii) When the switch is closed, there is an electric current flowing. How could you show that a magnet has been formed?
- (6)
- (iii) Give one use of an electromagnet.
- (3)
- (iv) Give one advantage of an electromagnet over a permanent magnet.
- (3)
- (c) Describe an experiment to investigate the release of stored energy from food. (9)

