AN ROINN OIDEACHAIS

(Department of Education).

BRAINNSE AN MHEADHON-OIDEACHAIS (Secondary Education Branch).

INTERMEDIATE CERTIFICATE EXAMINATION, 1937

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FULL COURSE. SCIENCE (Syllabus A)

FRIDAY, 18th JUNE.—AFTERNOON, 4 TO 6 P.M.

[Not more than six questions to be attempted. All questions carry equal marks. Illustrate your answers wherever possible.]

1. State Archimedes' Principle.

How would you verify it experimentally?

A solid weighs 19.8 grm. in air, 17.9 grm, in water, and 18.2 grm. in turpentine. Calculate the specific gravity of turpentine.

2. Describe, with sketch, how you would suspend a simple pendulum, and how you would determine its time of swing.

How would you measure the length of the pendulum? How is the time of swing affected by—

- (a) the length of the pendulum;
- (b) the mass of the bob;
- (c) the angle through which the pendulum swings?
- 3. The following table was obtained by experiments on circular discs cut from a sheet of metal.

Weight of disc in grams.	1.8	6.2	9.2	17-1	24.8	40.7
Diameter in cms	1.1	2.0	2.45	3.3	4.0	5.1

Show these results in the form of a graph and deduce the relation between the weight of a disc and its diameter.

4. What do you understand by "standard temperature and pressure"?

State the laws connecting the volume of a given mass of dry gas with (a) its pressure, (b) its temperature.

Name the discoverer of each law.

A certain mass of gas occupies 500 c.c. at 800 mm. pressure and 15°C. Determine its volume at standard temperature and pressure.

5. Explain the terms :-

(a) real expansion of a liquid;

(b) apparent expansion of a liquid.

How would you determine experimentally the coefficient of apparent expansion of a liquid? Calculate the coefficient of expansion of mercury-in-glass from the following figures:—

Mass of glass bottle=30.20 grm.

Mass of glass bottle filled with mercury at 15°C.=430·20 grm. Mass of glass bottle filled with mercury at 100°C.=424·97 grm.

6. Define "Latent Heat of Fusion."

A copper calorimeter (specific heat 0·1) weighs 50 grm. It contains 65 grm. of water at 35°C. 10 grm. of dry ice at 0°C. are introduced. The final temperature is 20·6°C. Determine the Latent Heat of Fusion of ice.

How would you perform the above experiment?

What precautions are necessary to secure an accurate result?

7. What do you understand by the term "solubility of a salt"? Describe fully how you would determine the solubility of nitre at 50°C. How is the result calculated?

8. Define "chemical equivalent of an element." Draw a sketch of the apparatus you would use to determine the chemical equivalent of zinc.

How would you carry out the experiment? How is the result calculated?

9. What happens when :-

(a) limestone is strongly heated,

(b) steam is passed over heated iron,

(c) air is passed over heated copper,

(d) nitre is heated with concentrated sulphuric acid?

Name the products in each case. Sketch the apparatus you would use to collect the product in (d).

10. State the "Principle of Moments."

A uniform metre stick is supported at its ends from two vertical spring balances. A mass of 60 grm. hangs from the 30 cm. mark and a mass of 120 grm. hangs from the 70 cm. mark. If the mass of the stick is 90 grm, determine the reading of each spring balance.

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(a) the Triangle of Forces;

(b) the Parallelogram of Forces.

Describe in detail how you would verify (a) experimentally.

12. What do you understand by the term "work" in mechanics? How is work measured?

A mass of 20 lb. is pulled through a distance of 10 feet along a horizontal plane. If the coefficient of friction is 0.2, determine the work done.

If the plane is inclined at 30° to the horizontal, how much extra work would be done in pulling the same mass 10 feet up along the plane?